

Blood gas Workshop

VTA 2019

Acid

- A substance that dissociates in $\text{H}_2\text{O} \longrightarrow \text{H}^+$
- $\text{HCl} \longrightarrow \text{H}^+ + \text{Cl}^-$ (complete dissociation)
- Acetic acid $\xrightleftharpoons{\text{blue}} \text{H}^+ + \text{Acetate}^-$

Base

- Substance that accepts a H^+

Buffer

- Mixture of a weak acid and its conjugate base

Bicarbonate buffer system (ECF)

- $\text{H}_2\text{CO}_3 / \text{HCO}_3^-$
- **Strong acid** : $\text{HCO}_3^- + \text{H}^+ \longrightarrow \text{H}_2\text{CO}_3$
- **Strong Base**: $\text{H}_2\text{CO}_3 + \text{OH}^- \longrightarrow \text{H}_2\text{O}$
- pH – measure of $[\text{H}^+]$ or $\text{Log } 1/ [\text{H}^+]$
- H/H equation relates the pH of a buffer to the concentration of its buffer acid and base
$$\text{pH} = \text{pKa} + \log [\text{base}] / [\text{acid}]$$

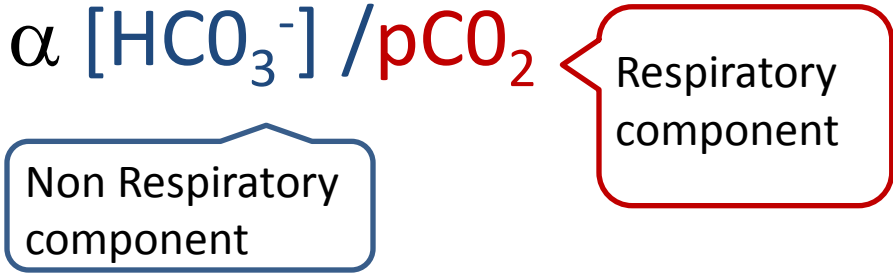
- $\text{pH} = \text{pK}_a + \log [\text{HCO}_3^-] / [\text{H}_2\text{CO}_3]$
- $\text{pH} = 6.1 + \log \frac{20}{1}$
 $= 7.4$

As long as the ratio 20:1 is kept, irrespective of concentrations, pH is normal

- $\text{H}_2\text{CO}_3 \rightleftharpoons \text{H}_2\text{O} + \text{CO}_2$
- $\text{H}_2\text{CO}_3 \xrightleftharpoons{\alpha} \text{pCO}_2$ } oversimplification

$$\text{pH} \propto [\text{HCO}_3^-] / \text{pCO}_2$$

Components of acid base status

- $\text{pH} \propto [\text{HCO}_3^-] / \text{pCO}_2$ 

Non Respiratory component

Respiratory component
- If disease causes changes in one component, the other component responds in an **attempt to correct** the ratio of 20: 1
- See if you can work out the direction of the compensatory component should one component change

Equations

- Metabolic acidosis
- Metabolic alkalosis
- Respiratory acidosis
- Respiratory alkalosis

Delta ratio – Increase in AG / Decrease in HCO_3

$$\text{AG calculated} - (12) / 24 - (\text{HCO}_3 \text{ measured})$$

Interpretation

<0.4 – Normal AG Met Acid

0.4 – 0.8 – Mixed Normal AG , High Metabolic Met Acid

1-2 - High AG met Acid

> 2 - HCO_3 is high to start with

Case 1

pH 7.08

PaCO₂ 18,8 mmHg

PaO₂ 100 mmHg

HCO₃ 5,4

Na 136 mmol/l

Cl 102 mmol/l

FiO₂ 0.3

Case 2

pH 7.48

PaCO₂ 47 mmHg

PaO₂ 68 mmHg

HCO₃ 34

FiO₂ 0.4

K = 2,8

Metabolic Alkalosis

- High pH
- HCO_3^- excess
- Compensation?

$$\text{pH} \propto [\text{HCO}_3^-] / \text{pCO}_2$$

Causes of Metabolic alkalosis

- H^+ loss
 - GIT:
 - Vomiting / NG suction
 - Renal:
 - Diuretics
 - Mineralocorticoid excess
 - Low PTH/
 - Post chronic hypercapnea
 - Hypokalemia
 - Transcellular shift/ Loss

- HCO_3^- excess
 - NaHCO_3 admin
 - Massive transfusion
- Contraction alkalosis

Maintenance of Metabolic alkalosis

- Hypovolemia

Proximal Tubule in H^+ Homeostasis

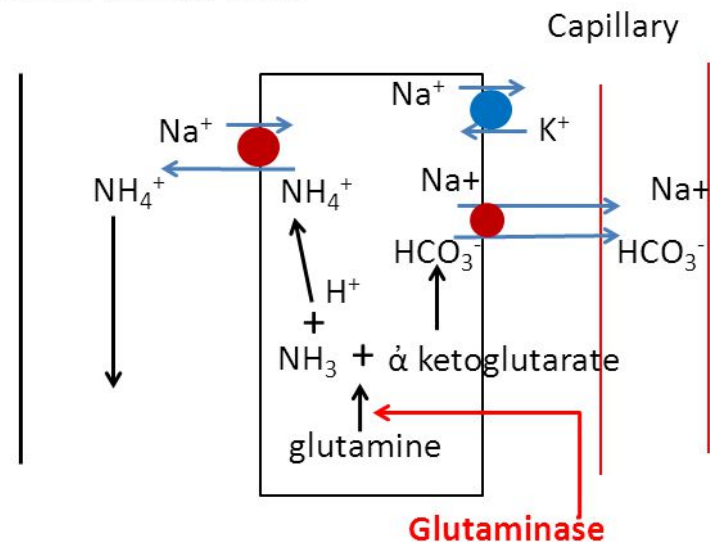
Low ECV

Increased Na reabsorption

Increased NH_4^+ excretion

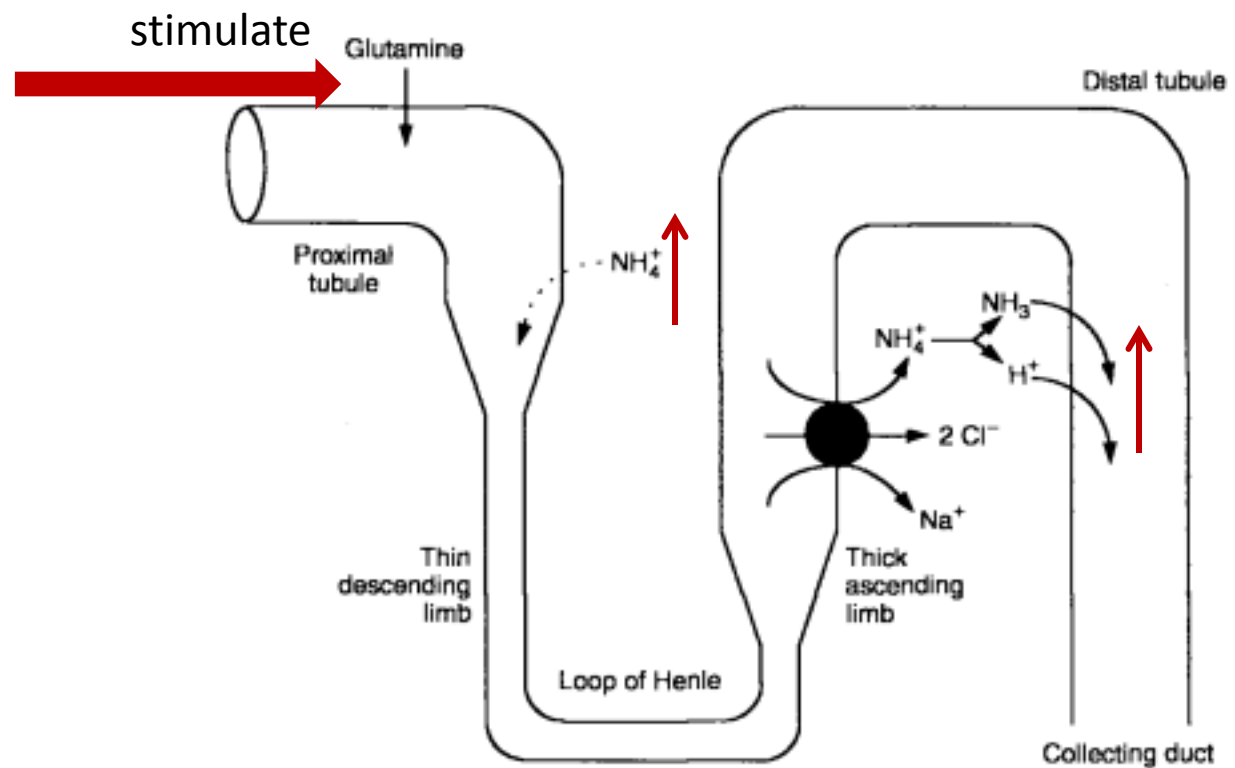
HCO_3^- reabsorption

Secretion of NH_4^+ by proximal tubular cells



Maintenance of Metabolic alkalosis

- Hypokalemia



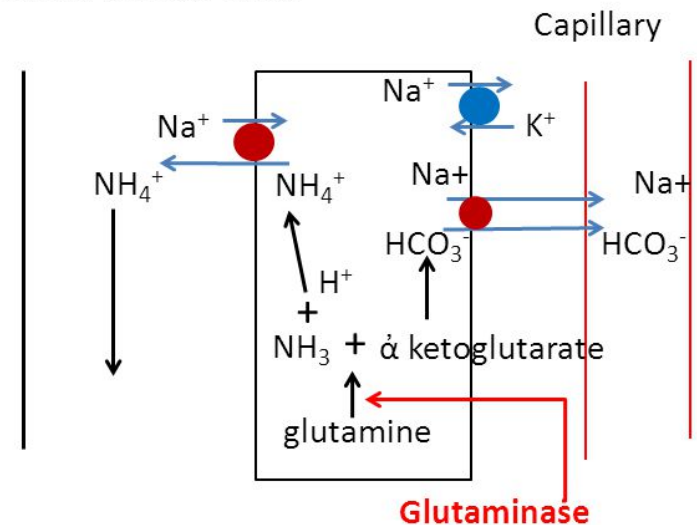
Management

1. Saline responsive

- Re-expand IVV and break cycle

Proximal Tubule in H^+ Homeostasis

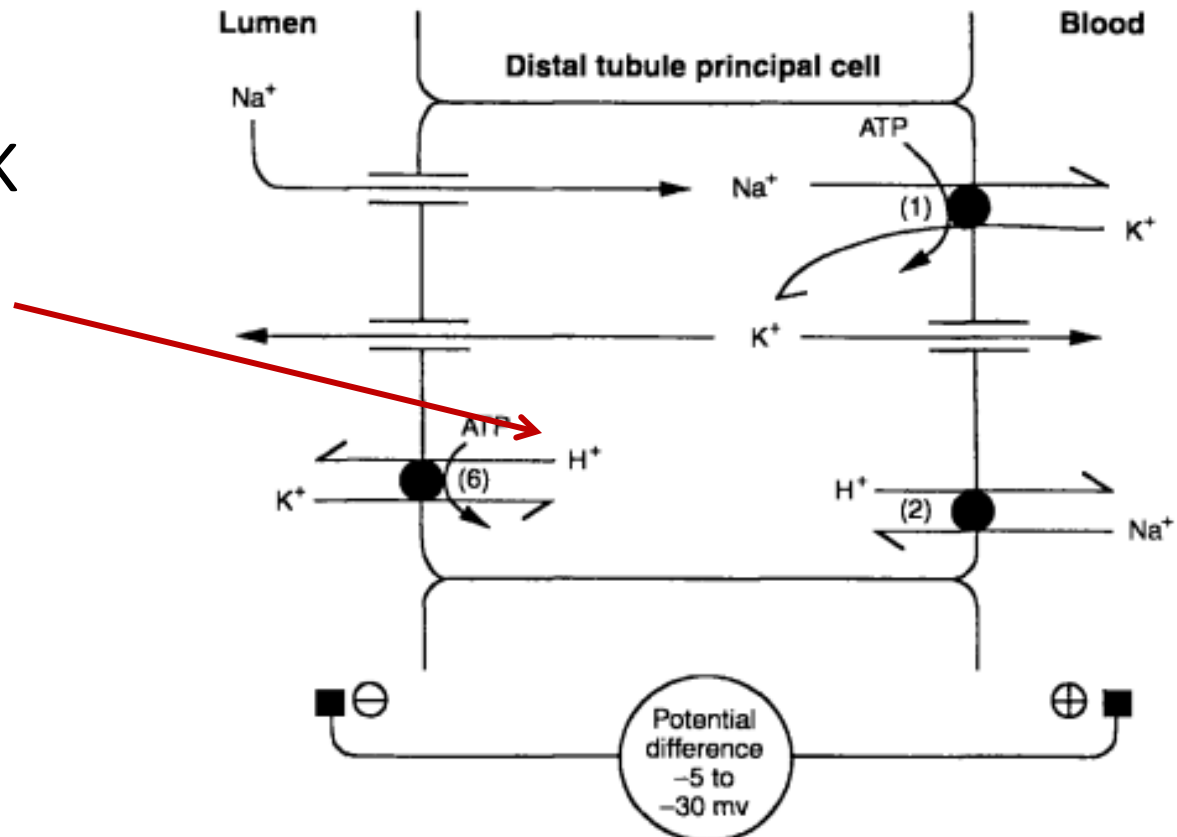
Secretion of NH_4^+ by proximal tubular cells



Management

2. Saline resistant

- Low K driving
- Plan- Correct K



Case 3

pH 7.50

PaCO₂ 41 mmHg

PaO₂ 145 mmHg

HCO₃ 34

FiO₂ 0.4

K = 2,8

Causes:

- 1.vomiting in pregnant patient,
- 2.diuretics or vomiting in patient with chronic respiratory alkalosis typical of cirrhosis,
3. postcardiac arrest (hyperventilation, bicarbonate therapy and conversion of lactate to bicarbonate

Case 4

- Elderly COPD
- pH 7.55
- PaCO₂ 53 mmHg
- PaO₂ 63 mmHg
- Bicarbonate 48 mmol/l
- FiO₂ 0.21

Alveolar arterial gradient

Hypoxemia $\text{PaO}_2 < 80 \text{ mmHg}$

Due to imbalance between:

Pulmonary ventilation and Pulmonary capillary blood flow

$$\text{Alveolar O}_2 = \text{FiO}_2 \times (\text{Atm Pressure} - \text{H}_2\text{O}) - \text{PaCO}_2 / \text{RQ}$$



Decrease if : PaCO_2 increases

$$\text{PaCO}_2 = \text{PaCO}_2 \text{ production} / (\text{RR} \times (\text{Vt} - \text{Vd}))$$

Case 5

- pH 7.44
- PaCO₂ 63 mmHg
- PaO₂ 52 mmHg
- Bicarbonate 42 mmol/l
- FiO₂ 0.28

A-aDO₂

- A-a DO₂ in hypoxemia

1. High:

- Diffusion/shunt/dead space

2. Normal: Altitude/FiO₂ low or PaCO₂ high

$$\text{Alveolar O}_2 = \text{FiO}_2 \times (\text{Atm Pressure} - \text{H}_2\text{O}) - \text{PaCO}_2 / \text{RQ}$$

- Resp depressants/ sedatives/ms relaxants/NM ds/Skeletal abnormalities

Paeds BASIC ICU Course

- Email:
- paedsbasicjhb@gmail.com

Is Balanced really the solution?

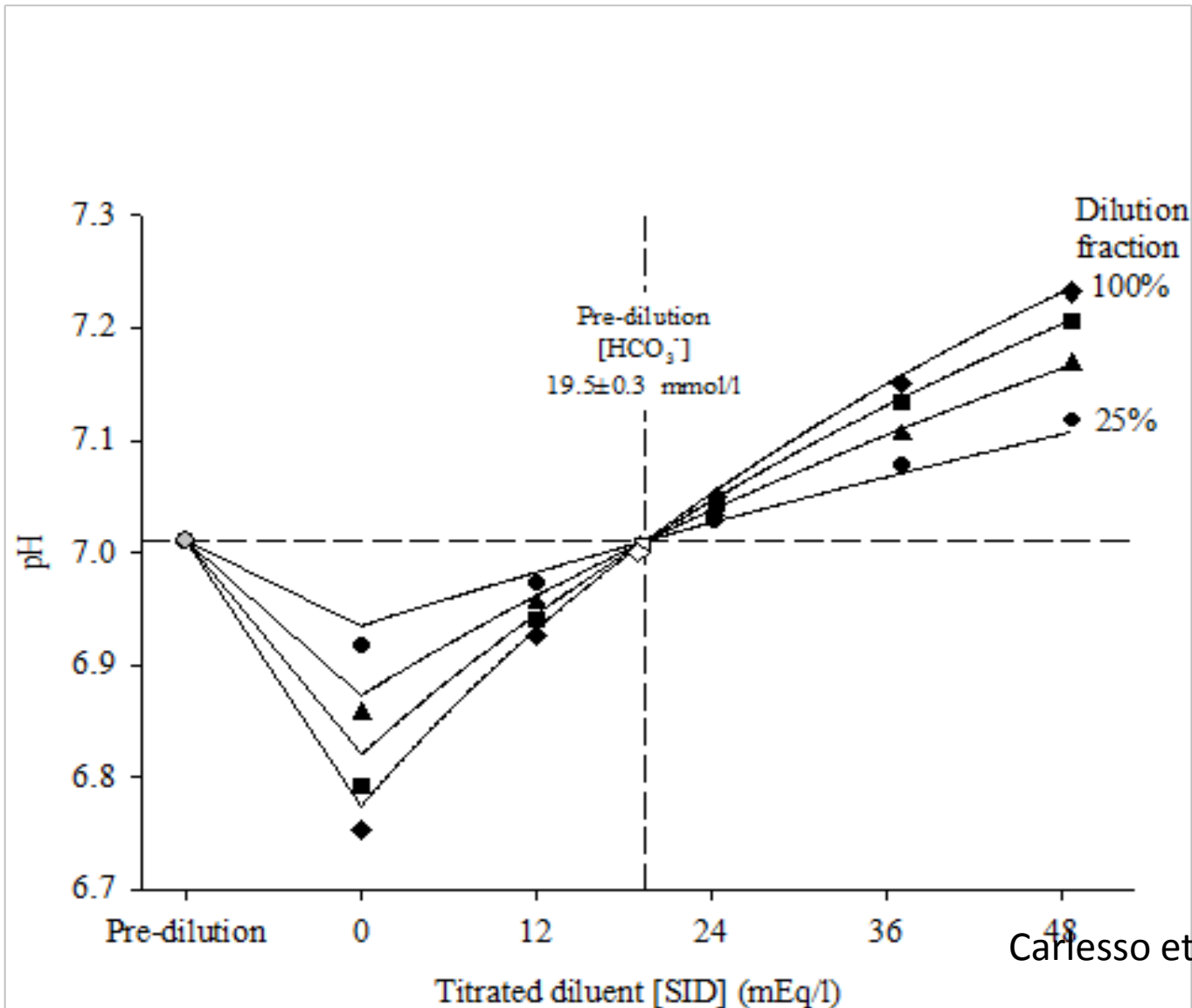
Shahed Omar

University of Witwatersrand/CHBAH

What is balanced?

Content	Plasma	Sodium chloride 0.9%*	Lactated Ringer's (USP)	Ringer's acetate	Alternative balanced solutions for resuscitation**	Bara'lyte
SID (Na+K-Cl)	40	0	28	27	50	52
Na ⁺ (mmol/l)	135–145	154	130	130	140	142
Cl ⁻ (mmol/l)	95–105	154	109	112	98	99
[Na ⁺]:[Cl ⁻] ratio	1.28–1.45:1	1:1	1.19:1	1.16:1	1.43:1	1.43:1
K ⁺ (mmol/l)	3.5–5.3	*	4	5	5	3.6
HCO ₃ ⁻ / Bicarbonate	24–32	0	28 (lactate)	27 (acetate)	27 (acetate) 23 (gluconate)	49
Ca ²⁺ (mmol/l)	2.2–2.6	0	1.4	1	0	0
Mg ²⁺ (mmol/l)	0.8–1.2	0	0	1	1.5	2.7
Glucose (mmol/ l)	3.5–5.5	0	0	0	0	0
pH	7.35–7.45	4.5–7.0	6–7.5	6–8	4.0–8.0	
Osmolarity (mOsm/l)	275–295	308	273	276	295	296

Balanced- depends on what you want

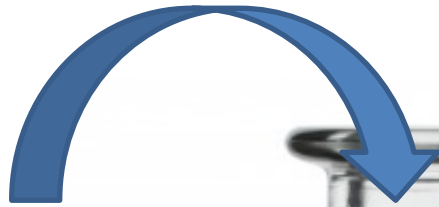


Respiration Physiology (1978) **33**, 9–26
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INDEPENDENT AND DEPENDENT VARIABLES OF ACID–BASE CONTROL¹

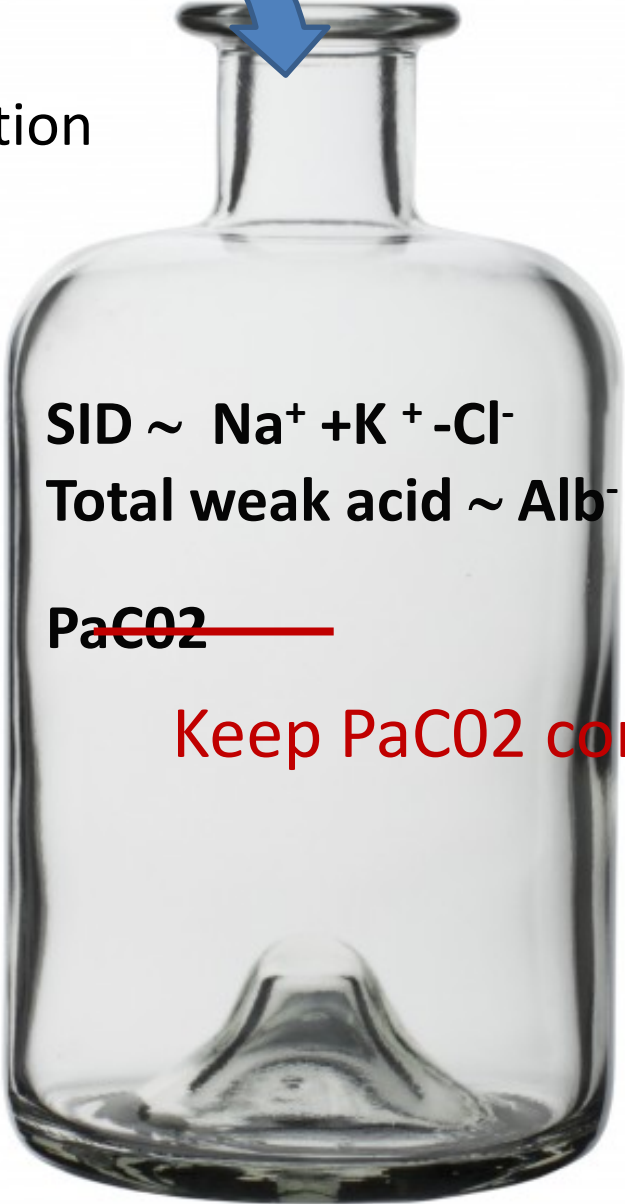
PETER A. STEWART

Physiology/Biophysics, Brown University, Providence, Rhode Island 02912, U.S.A.



Perfect solution

SID = 24



SID ~ $\text{Na}^+ + \text{K}^+ - \text{Cl}^-$
Total weak acid ~ Alb^-

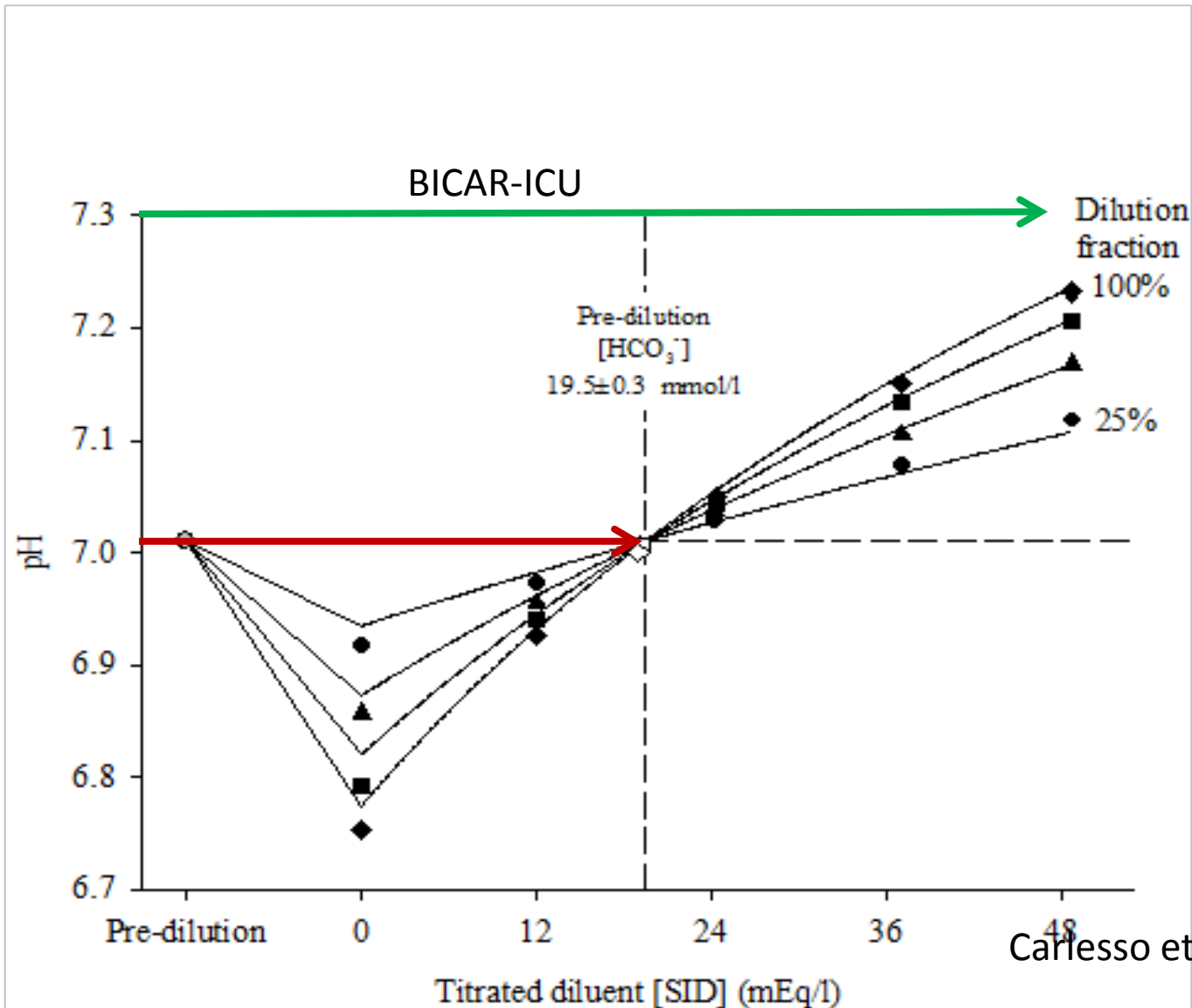
~~PaCO₂~~

Keep PaCO₂ constant

No effect on pH

Reduction the SID - $\uparrow \text{H}^+$
=
Reduction the Total weak acid - $\downarrow \text{H}^+$

Balanced- depends on what you want



Sodium bicarbonate therapy for patients with severe metabolic acidaemia in the intensive care unit (BICAR-ICU): a multicentre, open-label, randomised controlled, phase 3 trial



*Samir Jaber, Catherine Paugam, Emmanuel Futier, Jean-Yves Lefrant, Sigismond Lasocki, Thomas Lescot, Julien Pottecher, Alexandre Demoule, Martine Ferrandière, Karim Asehnoune, Jean Dellamonica, Lionel Velly, Paër-Sélim Abback, Audrey de Jong, Vincent Brunot, Fouad Belafia, Antoine Roquilly, Gérald Chanques, Laurent Muller, Jean-Michel Constantin, Helena Bertet, Kada Klouche, Nicolas Molinari, Boris Jung, for the BICAR-ICU Study Group**

Hypothesized

Early NaHCO₃ infusion vs no NaHCO₃ would result the composite of fewer deaths (D28) and SOFA 1by D7

- Inclusion (All)
 - $\geq 18y$
 - Within 48h of ICU admission
 - $pH \leq 7.2$ and $PaCO_2 \leq 45$
 - $SOFA \geq 4$ or $Lac \geq 2$
- Exclusion
 - Resp Acidosis/ HCO_3 loss from GIT or kidney
 - CKD – 4
 - Ketoacidosis
 - $NaHCO_3$ Rx or RRT within 24 of screening

942 patients were assessed for eligibility

542 excluded

109 already received sodium bicarbonate

87 were in terminal decline

76 had treatment limitation

69 had chronic renal failure

47 had immediate RRT indication

41 had ketoacidosis

37 had digestive loss of sodium bicarbonate

21 were eligible but not enrolled

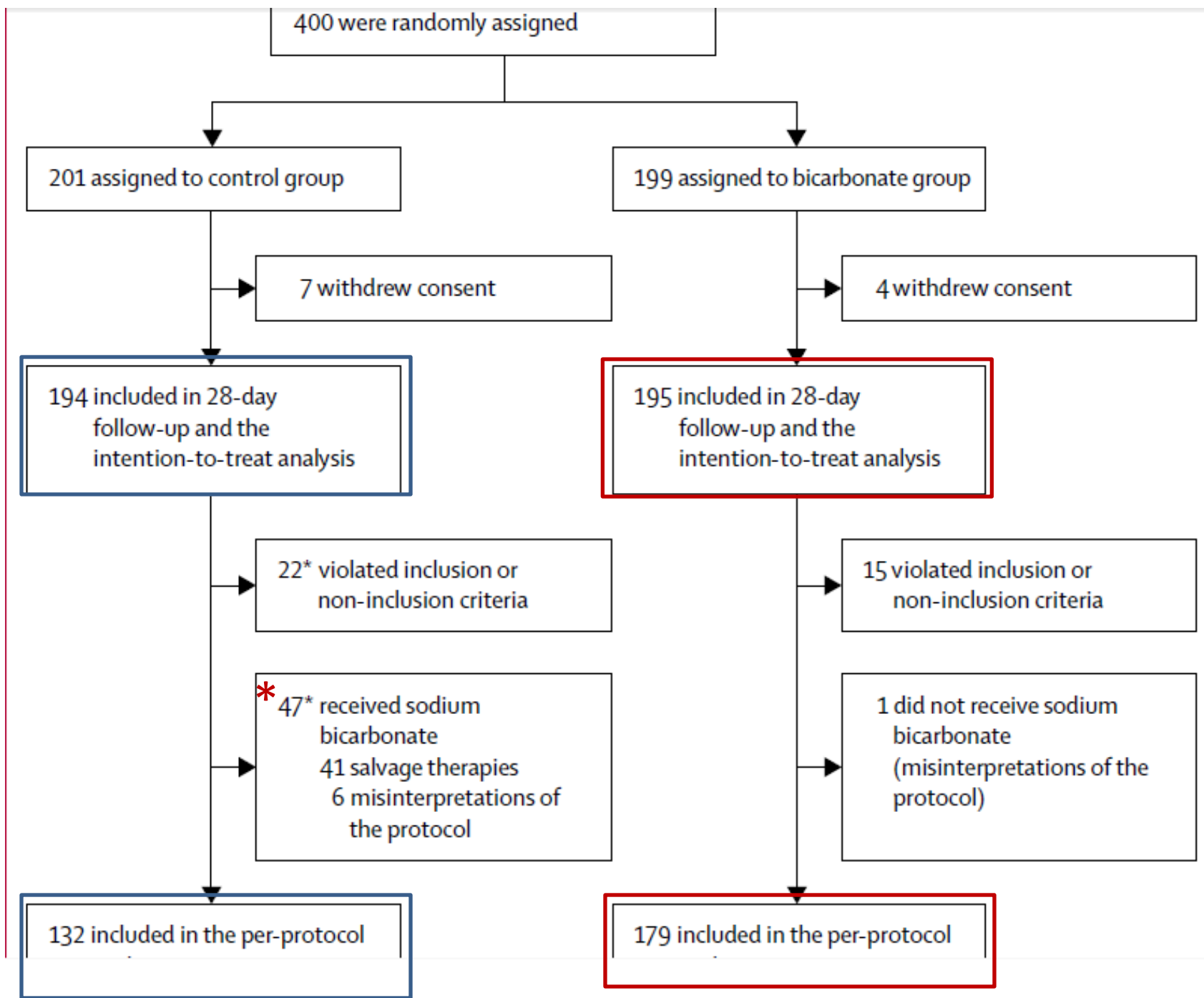
18 were included in another clinical study

13 had hyperkalaemia with heart signs

13 declined to participate

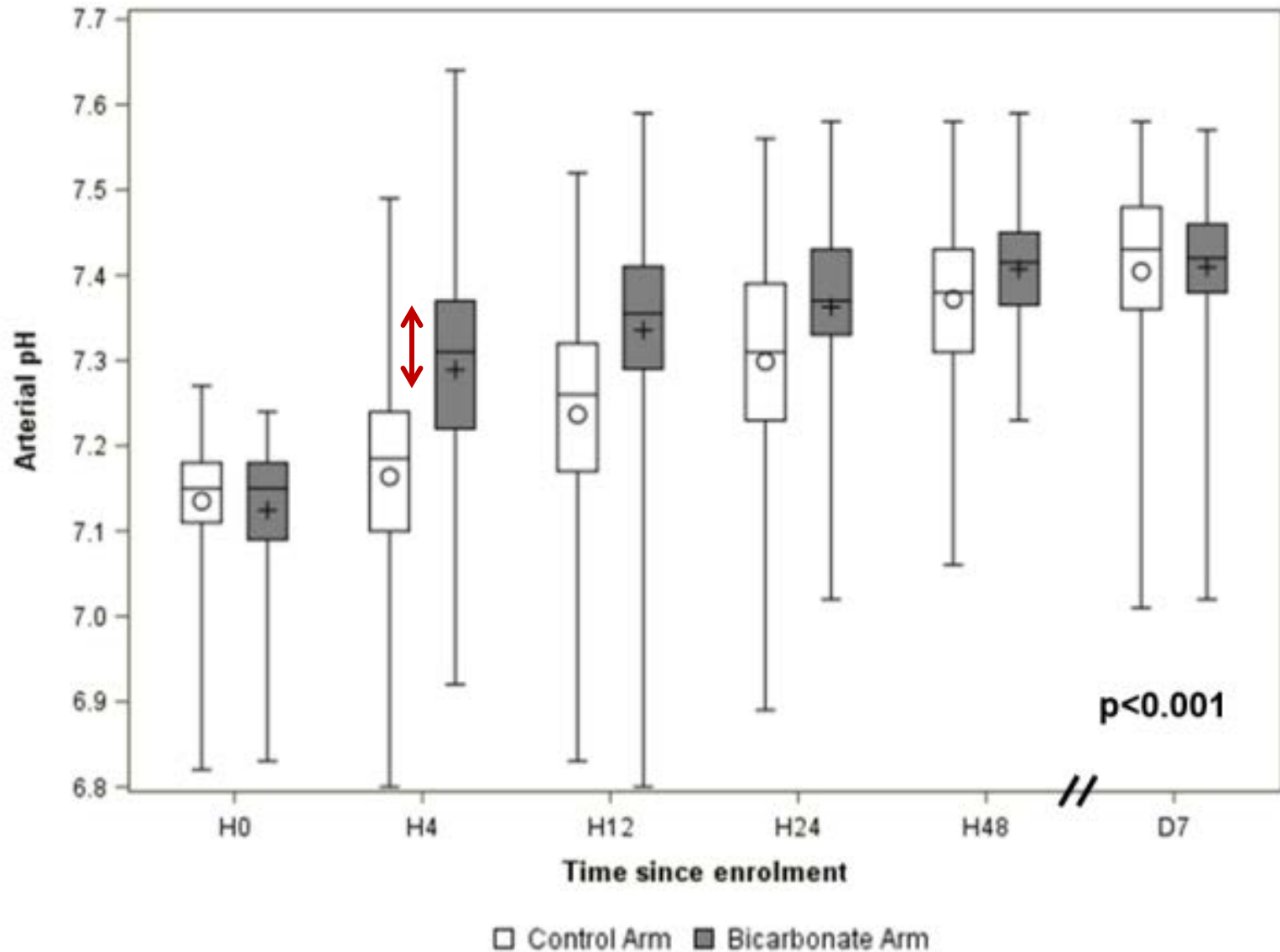
11 were under guardianship protection

400 were randomly assigned

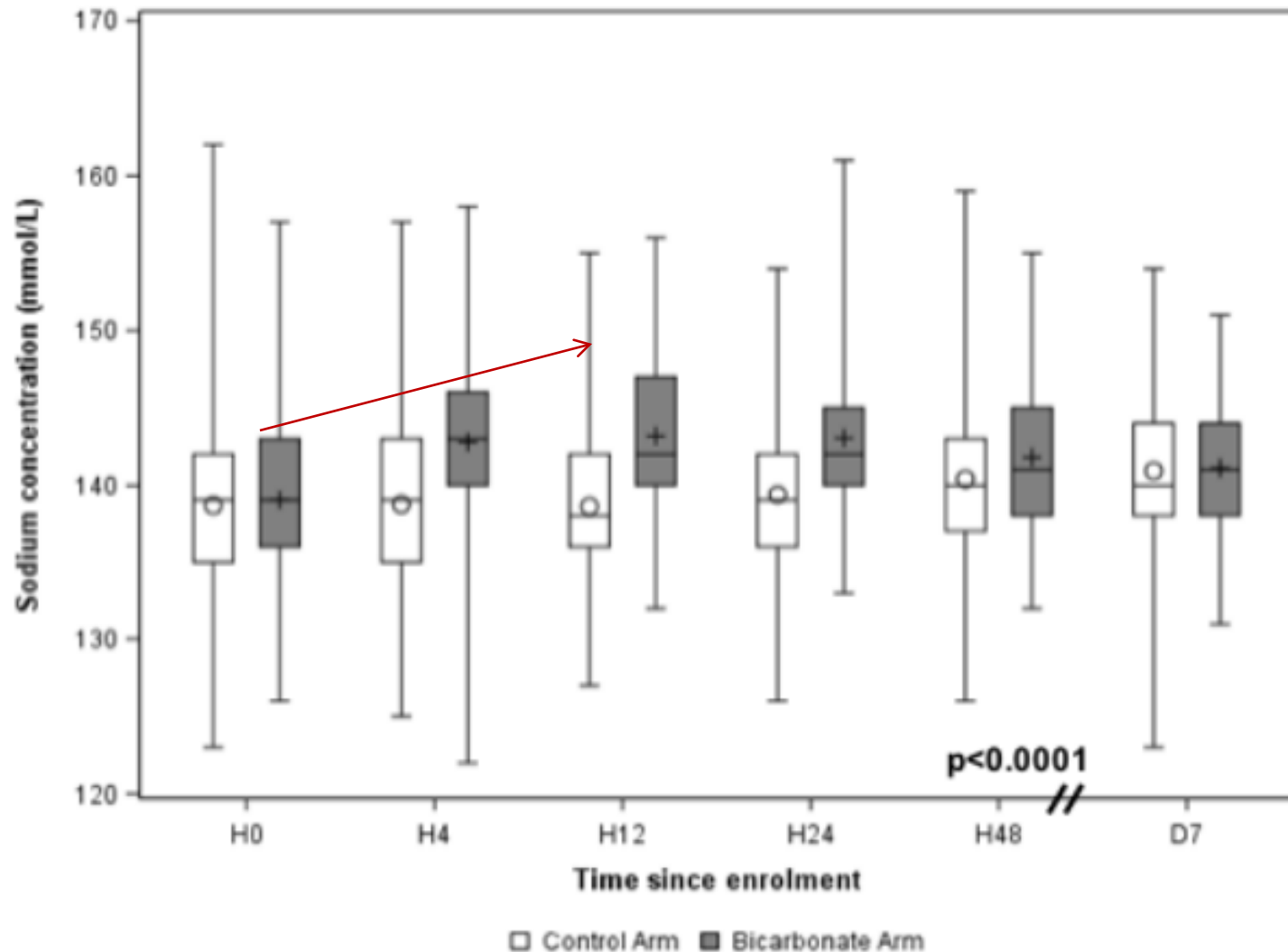


Other secondary outcomes	Control	Bicarb
Overall population (n=389)		
Cumulative fluid intake from enrolment to 24 h (mL)	3500 (1500–5250)	3350 (1800–5250)
Cumulative sodium bicarbonate volume intake from enrolment to 24 h (mL)	0 (0–0)	500 (250–750)
Cumulative sodium bicarbonate intake from enrolment to 24 h (mmol)	0 (0–0)	250 (125–375)
Cumulative fluid intake from 24 h to 48 h (mL)	1050 (0–2000)	1000 (0–2250)
Total Fluid 48h:	4550	4350
Na mmol/l	130	194
Cl	110	91
K	4	3.3
Est SID	24	~ 100

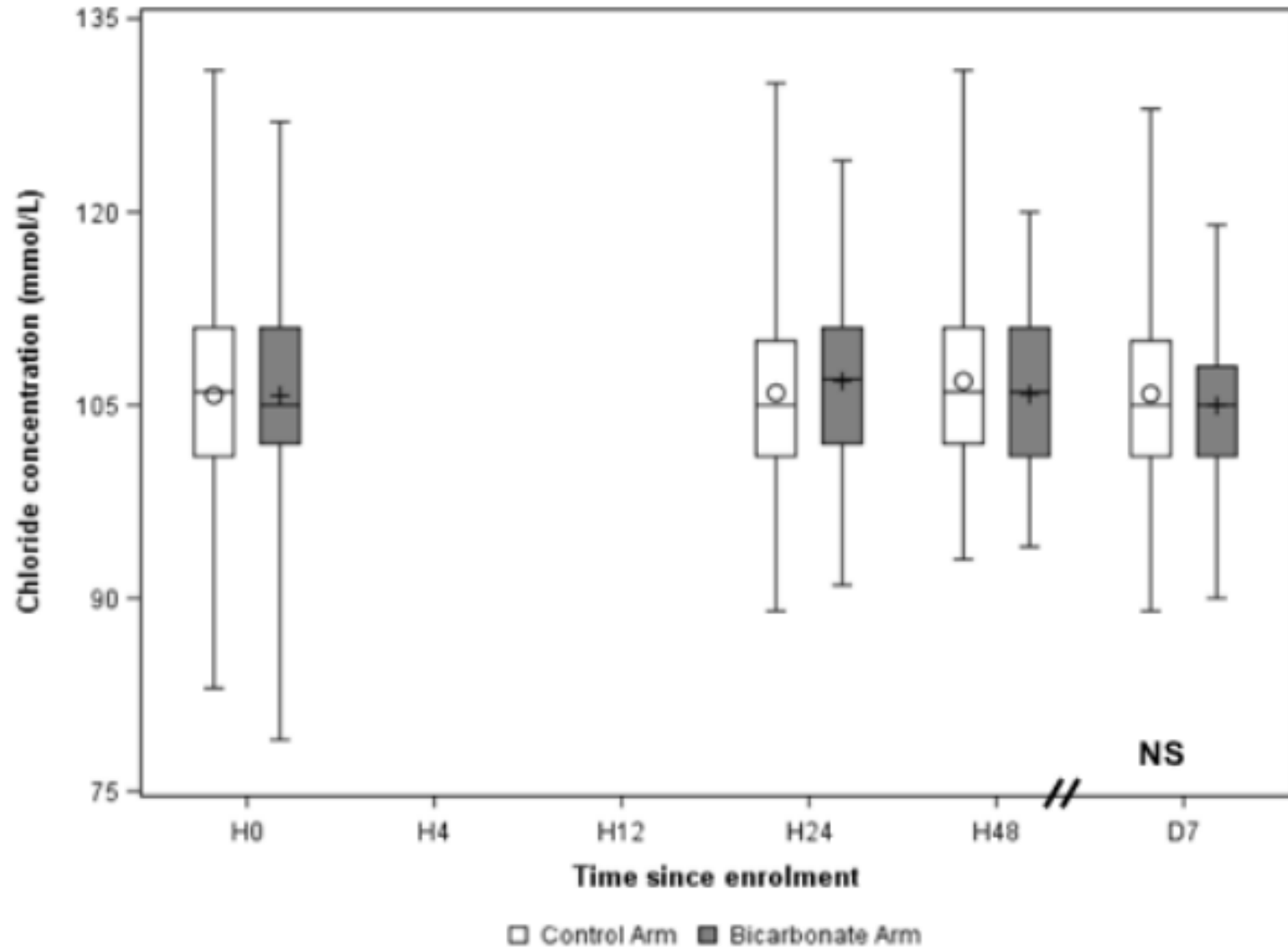
So, what happened to pH?



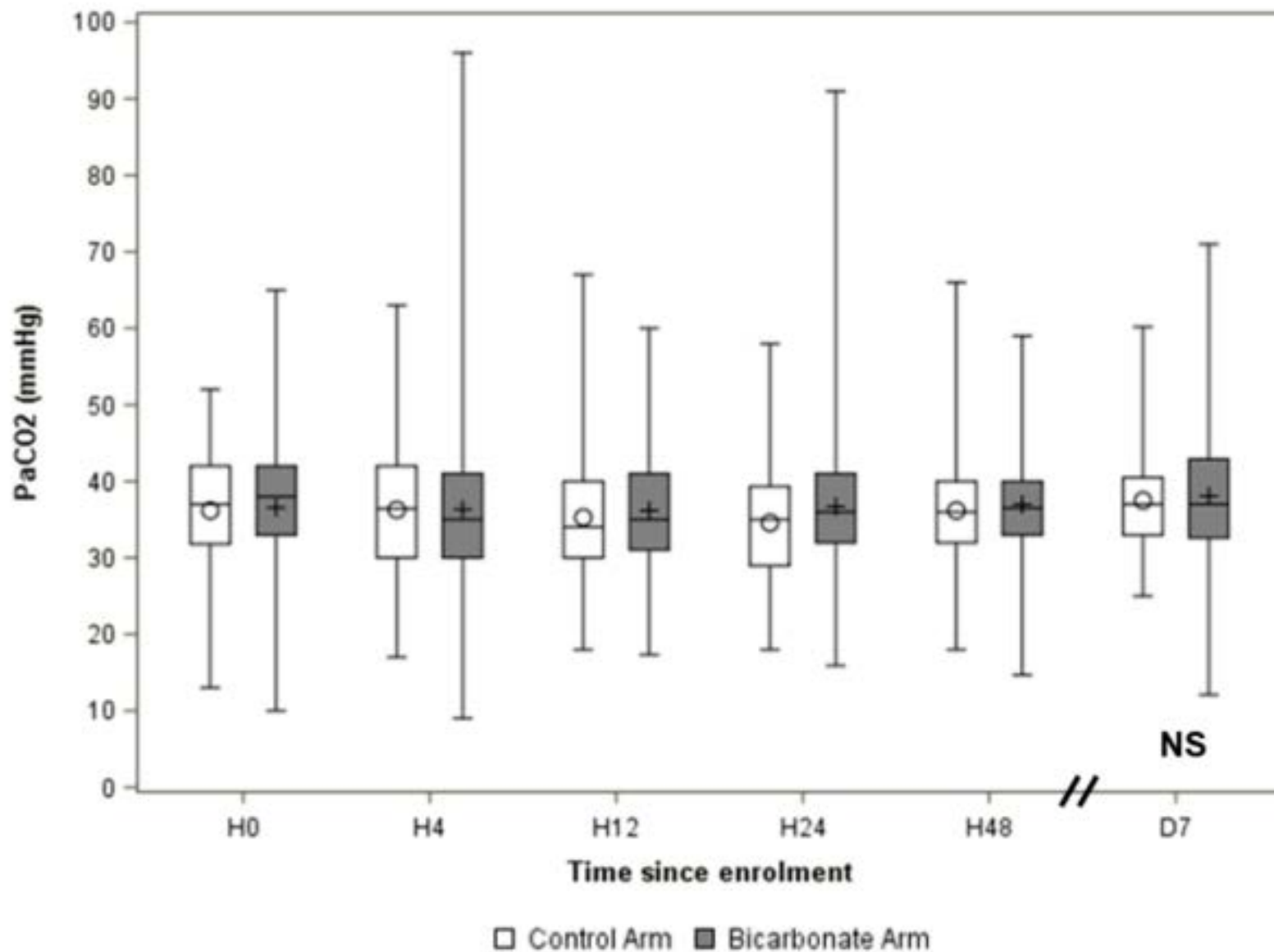
You can guess what happened to Na



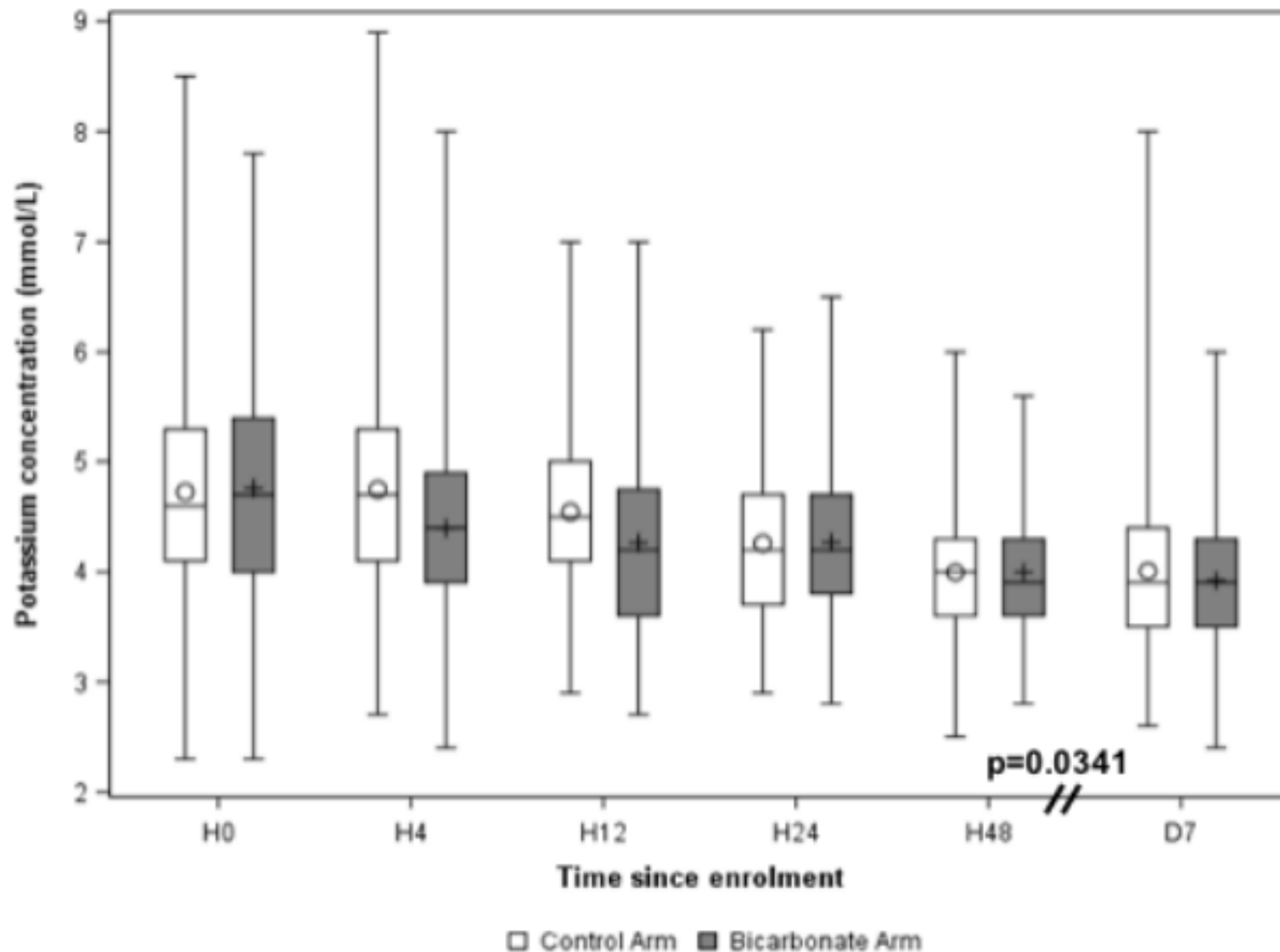
Chloride



What about PaCO₂ component?



Benefits of this strategy



Secondary outcomes

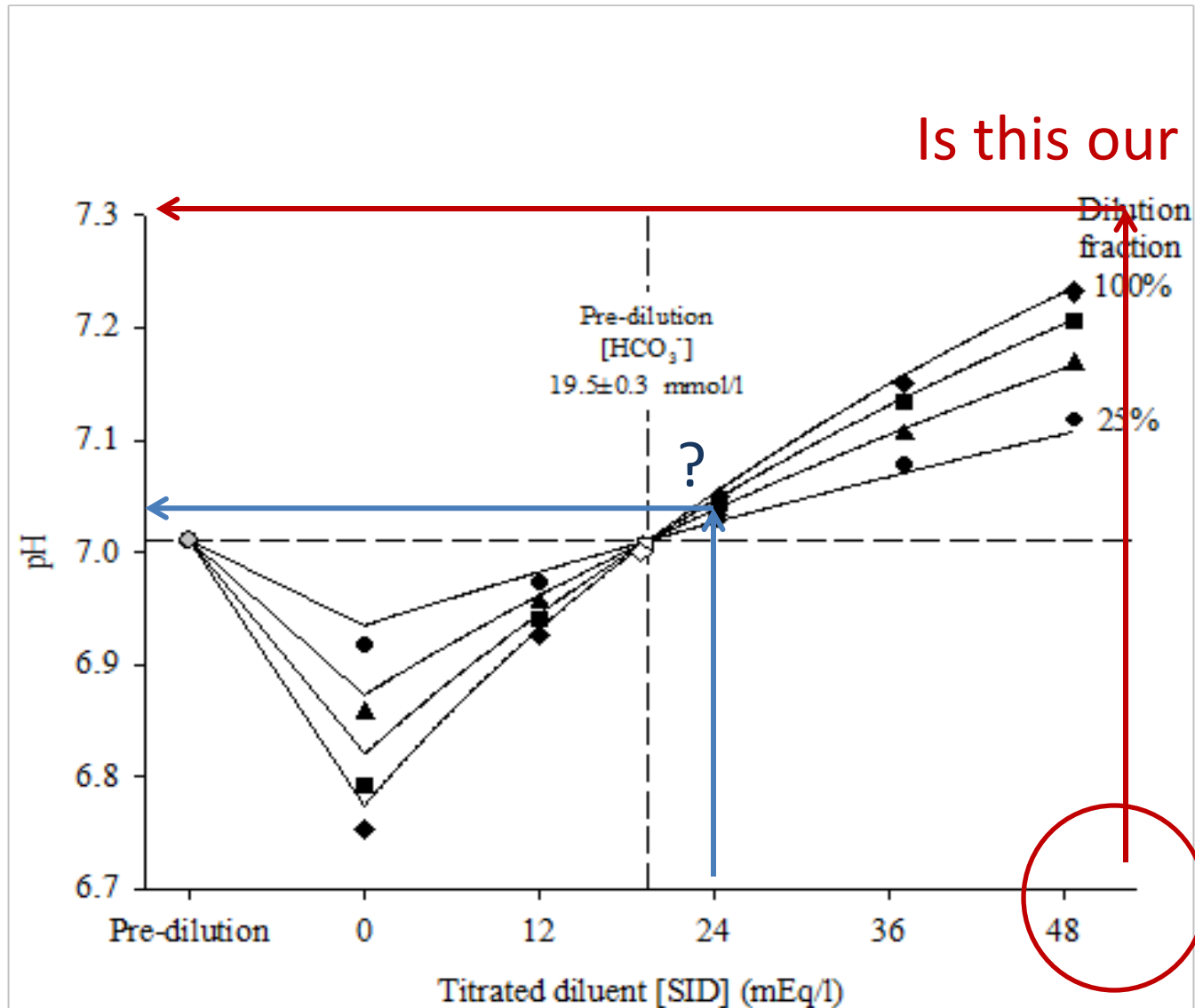
Renal replacement therapy

Overall population (n=389)

Use of renal replacement therapy during ICU stay	100 (52%)	*	68 (35%)
Time from enrolment to initiation of renal replacement therapy (h)	7 (3-18)	*	19 (7-82)
Renal replacement therapy-free days during ICU stay	8 (0-28)	*	19 (1-28)
Renal replacement therapy-free days during ICU stay in survivors	28 (25-28)		28 (25-28)
Dependence on dialysis at ICU discharge	11/32 (34%)		7/32 (22%)
Patients with AKIN scores of 2-3* (n=182)			
Use of renal replacement therapy during ICU stay	66/90 (73%)	*	47/92 (51%)
Time from enrolment to initiation of renal replacement therapy (h)	7 (3-17)	*	20 (8-82)
Renal replacement therapy-free days during ICU stay	1 (0-22)	*	10 (1-28)
Renal replacement therapy-free days during ICU stay in survivors	24 (22-28)		28 (19-28)
Dependence on dialysis at ICU discharge	10/21 (48%)	*	5/25 (20%)

	Control group (n=194)	Bicarbonate group (n=195)	
Primary outcome			Sig p
Overall population (n=389)			
Composite outcome	138 (71%)	128 (66%)	
Day 28 mortality	104 (54%)	87 (45%)	
At least one organ failure at day 7	134 (69%)	121 (62%)	
Patients with AKIN scores of 2-3* (n=182)			
Composite outcome	74/90 (82%)	64/92 (70%)	0.0462
Day 28 mortality	57/90 (63%)	42/92 (46%)	0.0166
At least one organ failure at day 7	74/90 (82%)	61/92 (66%)	0.0142

So back to our question - what you want?



	Control group (n=194)	Bicarbonate group (n=195)
Age		
Median age (years)	65 (55–75)	66 (55–75)
≥65	100 (52%)	104 (53%)
<65	94 (48%)	104 (47%)
Sex		
Men	123 (63%)	115 (59%)
Women	71 (37%)	80 (41%)
Body-mass index (kg/m ²)	27 (23–30)	26 (23–29)
Simplified Acute Physiology Score II*	60 (48–73)	59 (49–73)
Sepsis	115 (59%)	123 (63%)
AKIN status¶		
AKIN 0–1	104 (54%)	103 (53%)
AKIN 2–3	90 (46%)	92 (47%)

	Control group (n=194)	Bicarbonate group (n=195)
(Continued from previous page)		
Physiological support†		
Invasive mechanical ventilation	160 (82%)	164 (84%)
Vasopressor support	156 (80%)	154 (79%)
Laboratory results		
Arterial pH	7.15 (7.11–7.18)	7.15 (7.09–7.18)
PaO ₂ -to-FiO ₂ ratio (mm Hg)	229 (142–355)	264 (144–403)
PaCO ₂ (mm Hg)	37 (32–42)	38 (33–42)
Serum bicarbonate (mmol/L)	13 (10–15)	13 (10–15)
Serum lactate (mmol/L)	5.3 (3.4–9.0)	6.3 (3.6–9.7)
Serum lactate ≥2 mmol/L at enrolment	152 (78%)	168 (86%)
Serum creatinine (mg/dL)	1.76 (1.21–2.48)	1.67 (1.11–2.33)

What does the data indicate?

Meta-analysis of high- *versus* low-chloride content in perioperative and critical care fluid resuscitation

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³Boston Strategic Partners, Boston, Massachusetts, ⁴Baxter Healthcare Corporation, Deerfield, Illinois, and ⁵Department of Anesthesiology, Vanderbilt University Medical Center, Nashville, Tennessee, USA

Correspondence to: Professor A. D. Shaw, Division of Cardiothoracic Anesthesiology, Vanderbilt University Medical Center, Nashville, Tennessee 37232–8274, USA (e-mail: andrew.shaw@vanderbilt.edu)

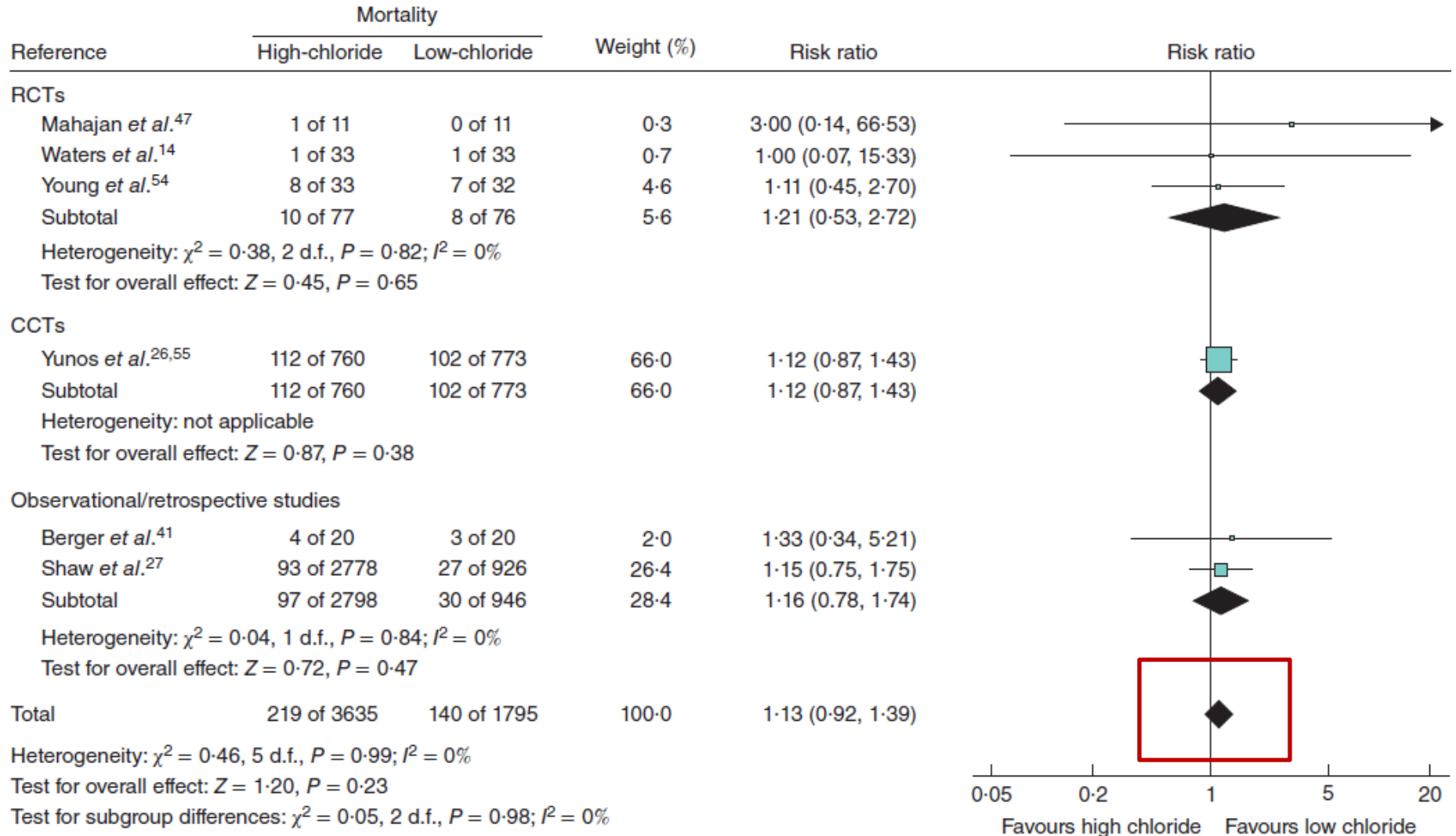
2014

Background: The objective of this systematic review and meta-analysis was to assess the relationship between the chloride content of intravenous resuscitation fluids and patient outcomes in the perioperative or intensive care setting.

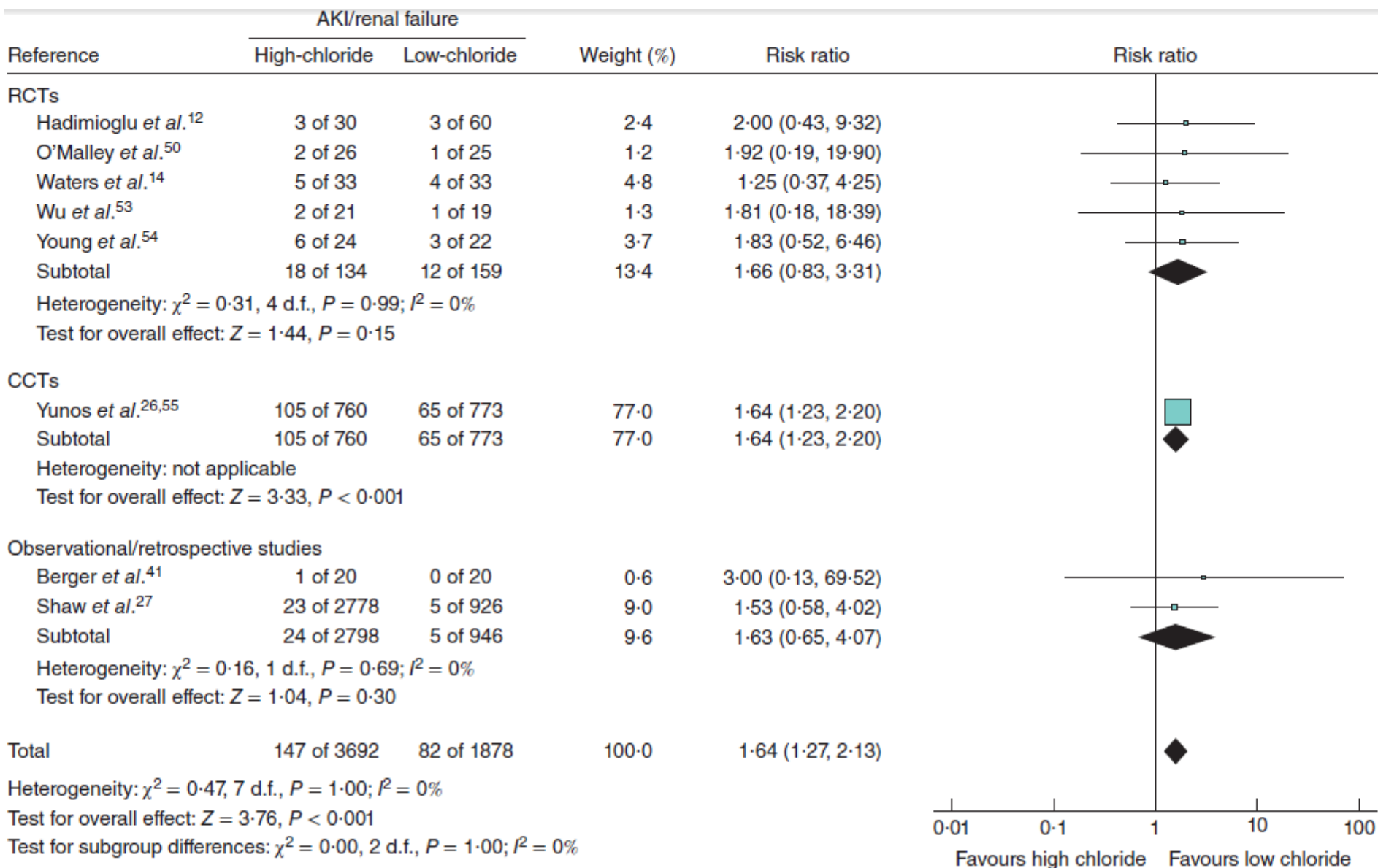
21 studies (6253 pt)

15 RCT, 5 Observational and 1 controlled clinical trial

Mortality



6 studies with mortality as an end point





However,

High Cl fluid:

Other

1. Metabolic acidosis (RR 2.87)
2. Greater serum Cl,
3. More Blood T/F
4. Longer MV

ORIGINAL ARTICLE

Balanced Crystalloids versus Saline in Critically Ill Adults

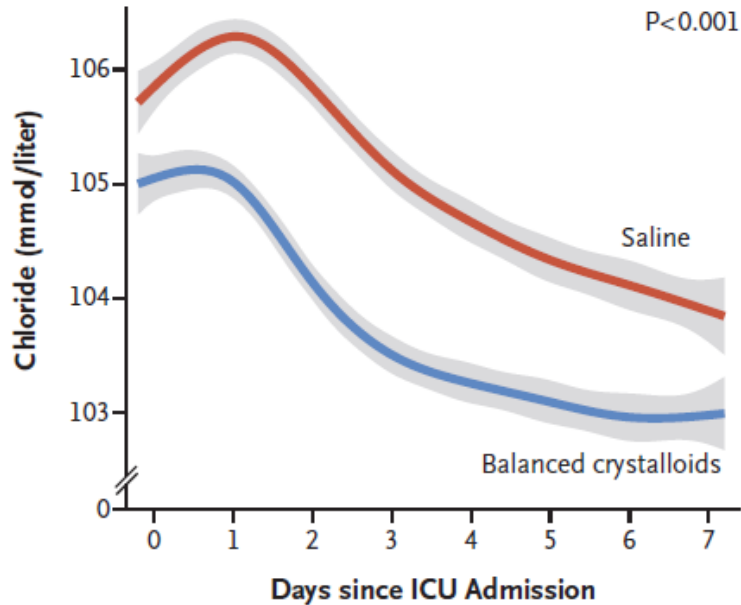
Matthew W. Semler, M.D., Wesley H. Self, M.D., M.P.H.,
Jonathan P. Wanderer, M.D., Jesse M. Ehrenfeld, M.D., M.P.H.,
Li Wang, M.S., Daniel W. Byrne, M.S., Joanna L. Stollings, Pharm.D.,
Avinash B. Kumar, M.D., Christopher G. Hughes, M.D.,
Antonio Hernandez, M.D., Oscar D. Guillaumondegui, M.D., M.P.H.,
Addison K. May, M.D., Liza Weavind, M.B., B.Ch., Jonathan D. Casey, M.D.,
Edward D. Siew, M.D., Andrew D. Shaw, M.B., Gordon R. Bernard, M.D.,
and Todd W. Rice, M.D., for the SMART Investigators
and the Pragmatic Critical Care Research Group* 2018

Pragmatic, unblinded, Cluster randomized, multiple crossover
Bal vs Saline, % ICU's in one Center, >15 000 patients

Outcomes

- 1° - MAKE 30 – Mortality/New RRT/Persistent renal dysfunction
- 2° Outcomes and Prespecified subgroup analysis

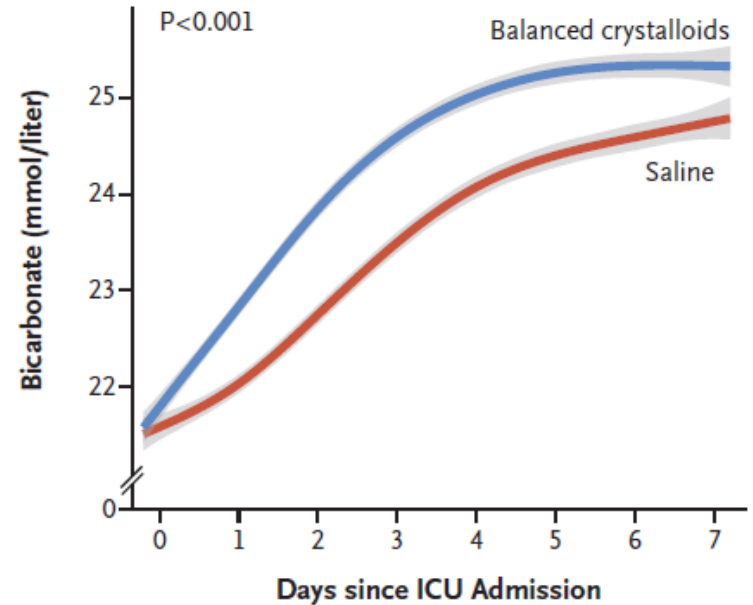
A Chloride Concentration



No. of Patients with Measurement

Balanced crystalloids	6904	4715	3263	2195
Saline	6747	4669	3283	2172

B Bicarbonate Concentration



No. of Patients with Measurement

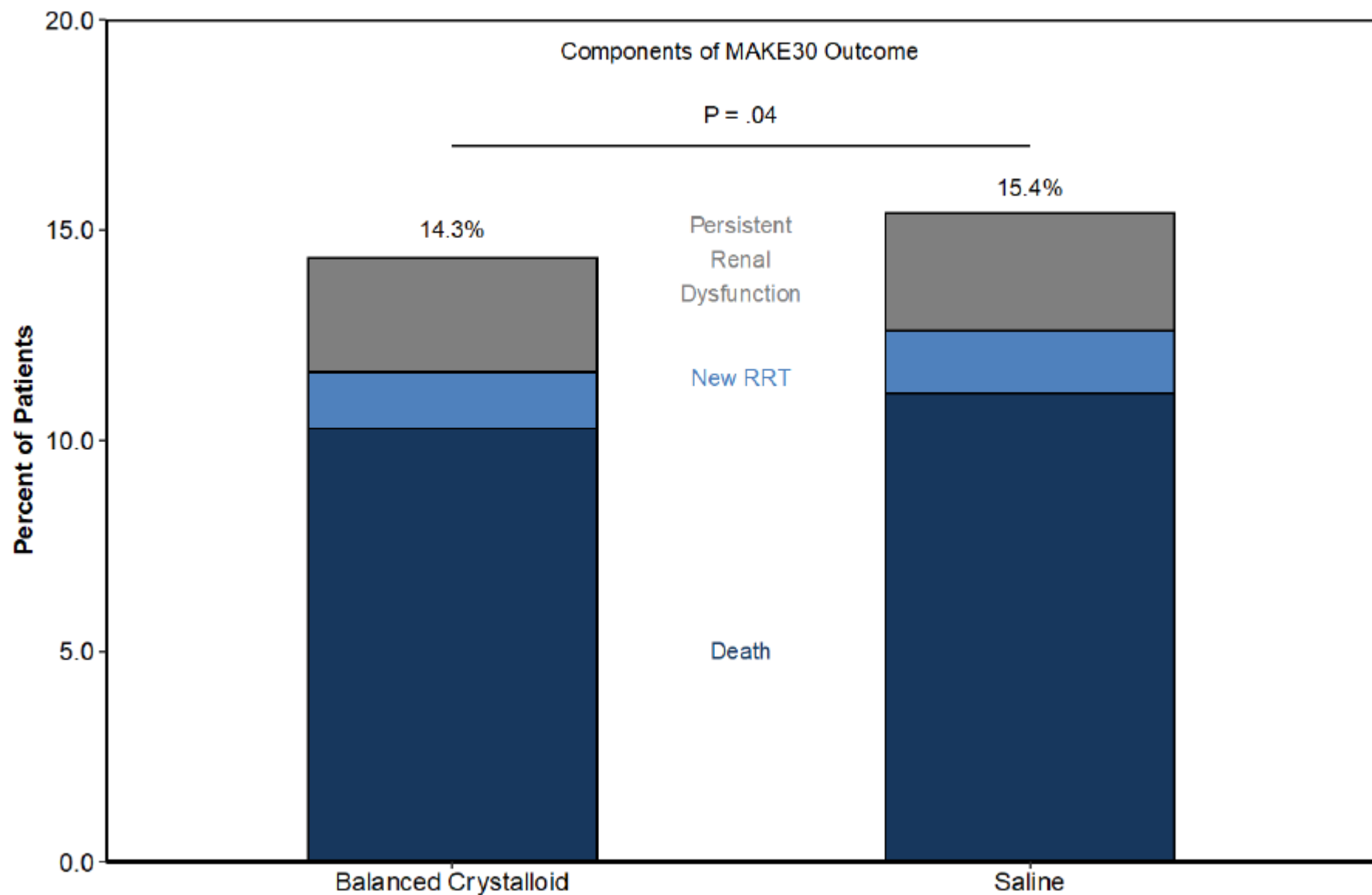
Balanced crystalloids	6929	4718	3266	2198
Saline	6763	4678	3293	2175

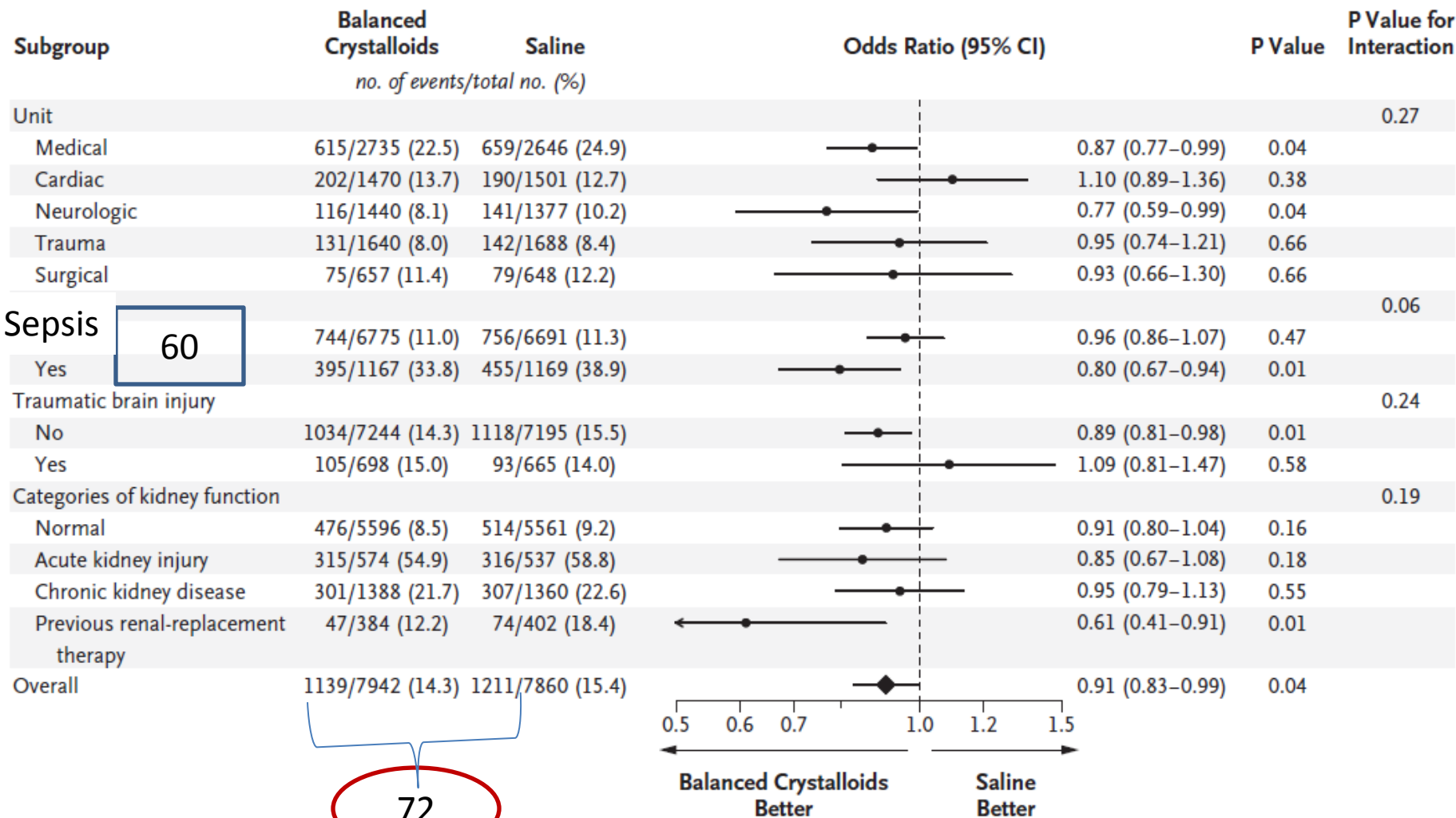
Figure 2. Plasma Chloride and Bicarbonate Concentration According to Group.

Due to co-ordination with ED and OR – Received same fluid pre-ICU, hence the differences at ICU admission

Table 2. Clinical Outcomes.*

Outcome	Balanced Crystalloids (N=7942)	Saline (N=7860)
Primary outcome		
Major adverse kidney event within 30 days — no. (%)‡	1139 (14.3)	1211 (15.4)
Components of primary outcome		
In-hospital death before 30 days — no. (%)	818 (10.3)	875 (11.1)
Receipt of new renal-replacement therapy — no./total no. (%)§	189/7558 (2.5)	220/7458 (2.9)
Among survivors	106/6787 (1.6)	117/6657 (1.8)
Final creatinine level $\geq 200\%$ of baseline — no./total no. (%)§	487/7558 (6.4)	494/7458 (6.6)



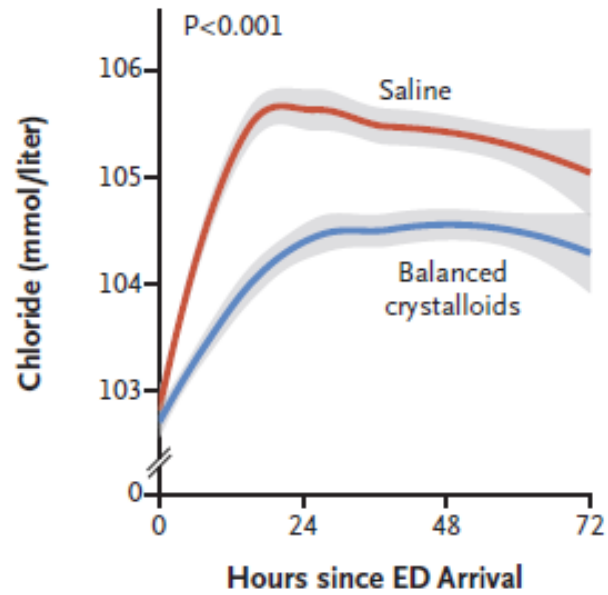


ORIGINAL ARTICLE

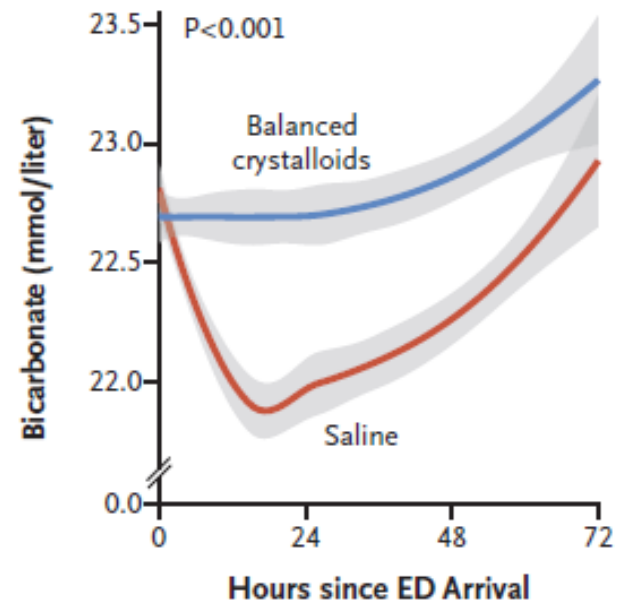
Balanced Crystalloids versus Saline in Noncritically Ill Adults

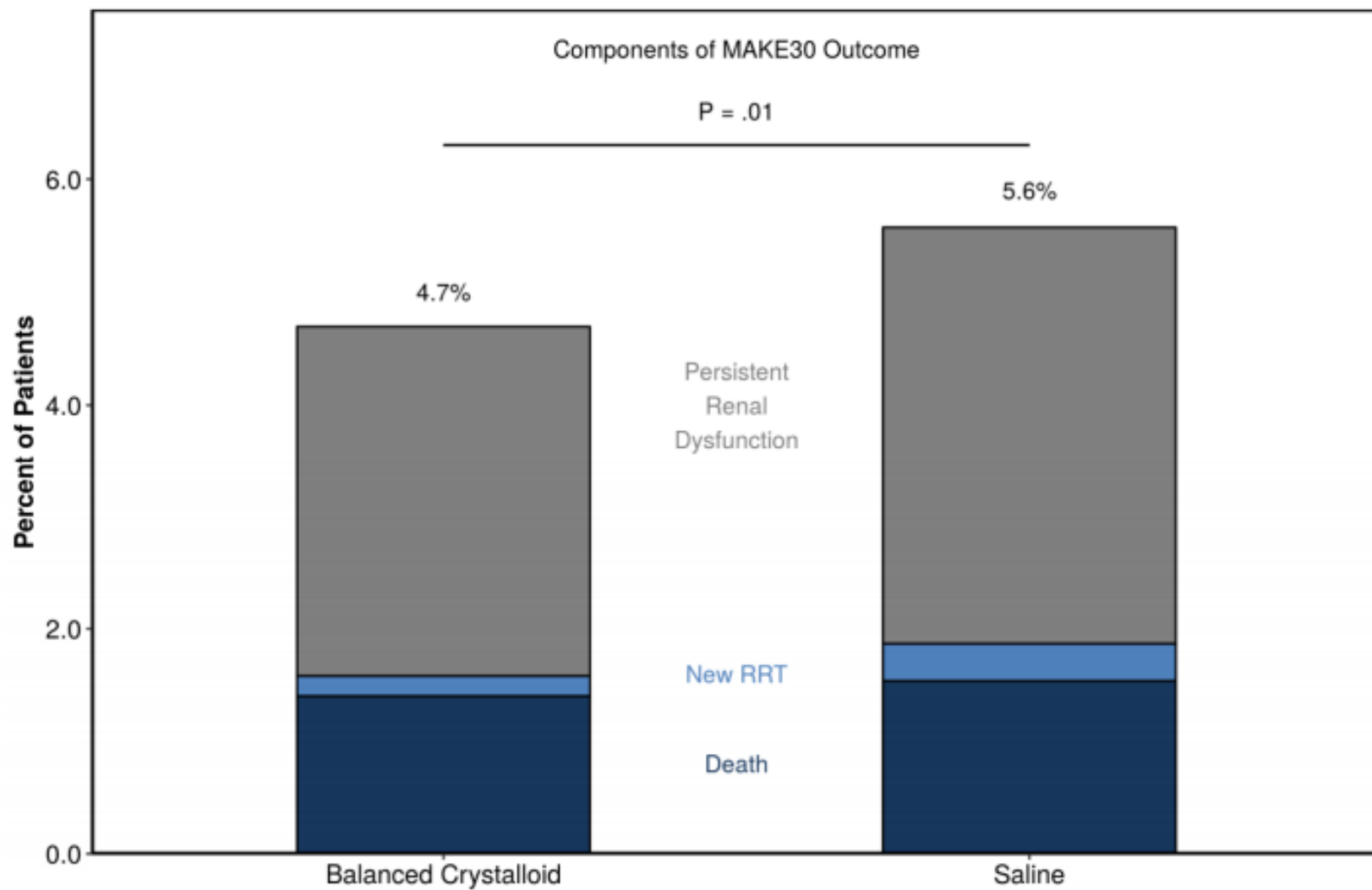
Wesley H. Self, M.D., M.P.H., Matthew W. Semler, M.D.,
Jonathan P. Wanderer, M.D., Li Wang, M.S., Daniel W. Byrne, M.S.,
Sean P. Collins, M.D., Corey M. Slovis, M.D., Christopher J. Lindsell, Ph.D.,
Jesse M. Ehrenfeld, M.D., M.P.H., Edward D. Siew, M.D.,
Andrew D. Shaw, M.B., Gordon R. Bernard, M.D.,
and Todd W. Rice, M.D., for the SALT-ED Investigators*

B Chloride

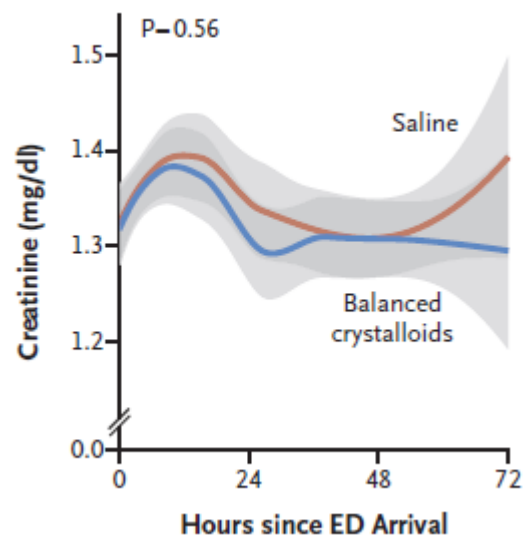


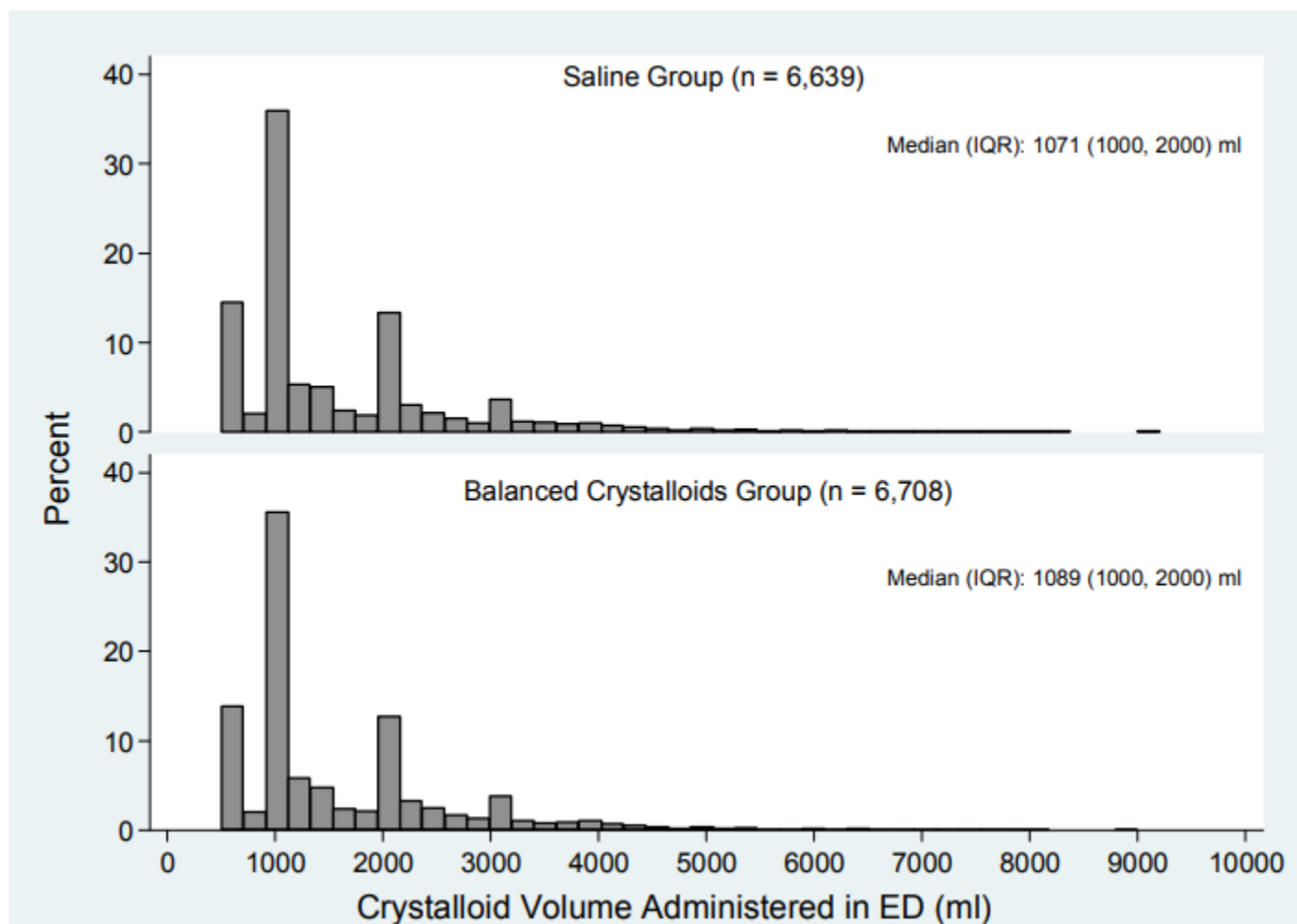
E Bicarbonate





F Creatinine






RESEARCH ARTICLE

Open Access



The prognostic importance of duration of AKI: a systematic review and meta-analysis

Swati Mehta^{1*} , Kinsuk Chauhan², Achint Patel², Shanti Patel², Rachel Pinotti², Girish N. Nadkarni², Chirag R. Parikh³ and Steven G. Coca²

- Duration of AKI was dependent on recovery sCr to within 25-50% of baseline sCr

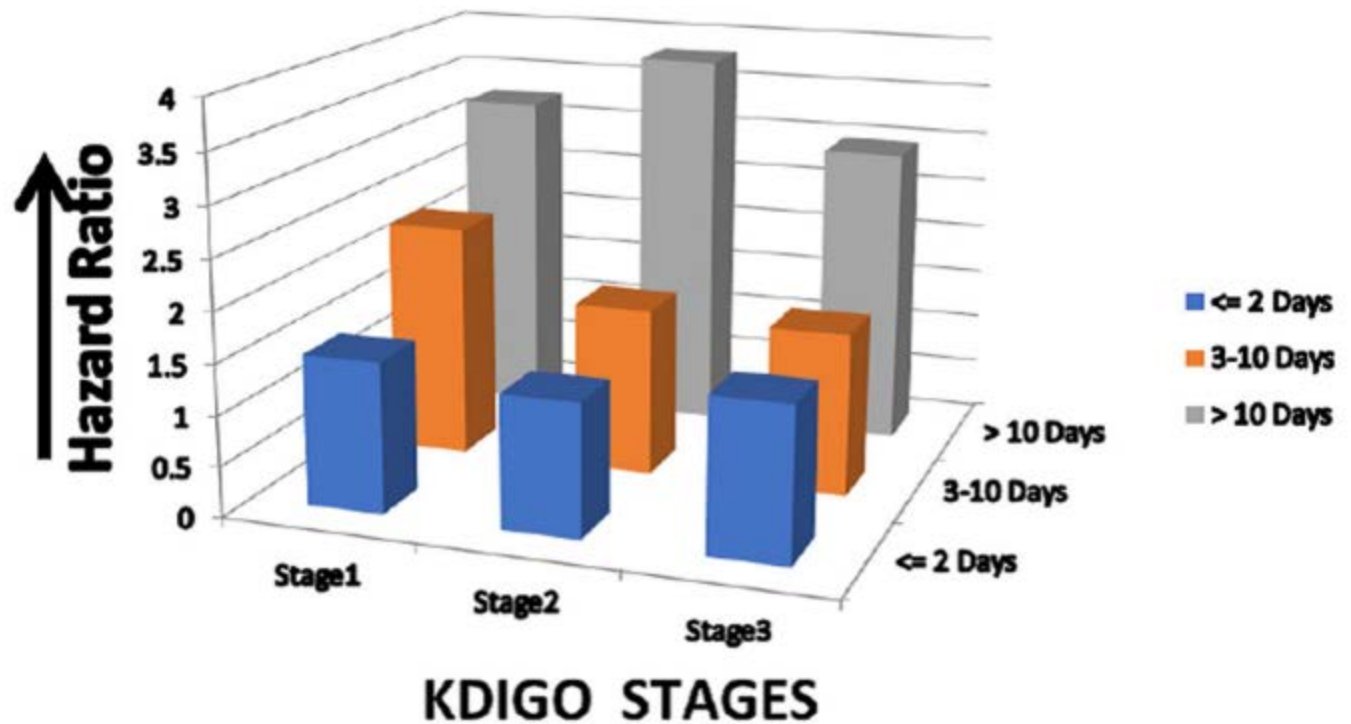


Fig. 5 Risk for incident CKD stage 3 increases with duration even after adjusting for KDIGO stage

Mehta S, Chauhan K, Patel A, Patel S, Pinotti R, Nadkarni GN, et al. The prognostic importance of duration of AKI: a systematic review and meta-analysis. BMC Nephrol [Internet]. 2018 Apr 19;19. Available from:

<https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5907696/>

Conclusion

- Duration of AKI is independently associated with long-term mortality and may provide additional prognostic information over and above magnitude of serum creatinine alone. Thus, AKI duration can be considered as a prognostic factor for long-term mortality and other cardiovascular outcomes and can be used as an endpoint in intervention trials to prevent or treat AKI.

Mehta S, Chauhan K, Patel A, Patel S, Pinotti R, Nadkarni GN, et al. The prognostic importance of duration of AKI: a systematic review and meta-analysis. BMC Nephrol [Internet]. 2018 Apr 19;19. Available from: <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC5907696/>

What about SPLIT

eTable 7. Cause-specific in hospital mortality within the 90-day follow-up period

	Buffered crystalloid group	Saline group	Relative Risk (95% CI)	P value
Categories – no. / total no. (%)				
Bleeding	3/1152 (0.3)	2/1110 (0.2)	1.45 (0.24 to 8.63)	1.00
Cardiac	14/1152 (1.2)	20/1110 (1.8)	0.67 (0.34 to 1.33)	0.30
Cerebral	32/1152 (2.8)	39/1110 (3.5)	0.79 (0.5 to 1.25)	0.34
Sepsis	30/1152 (2.6)	23/1110 (2.1)	1.26 (0.73 to 2.15)	0.41
Other	8/1152 (0.7)	11/1110 (1)	0.7 (0.28 to 1.74)	0.50

Total

87/1152
7.6%

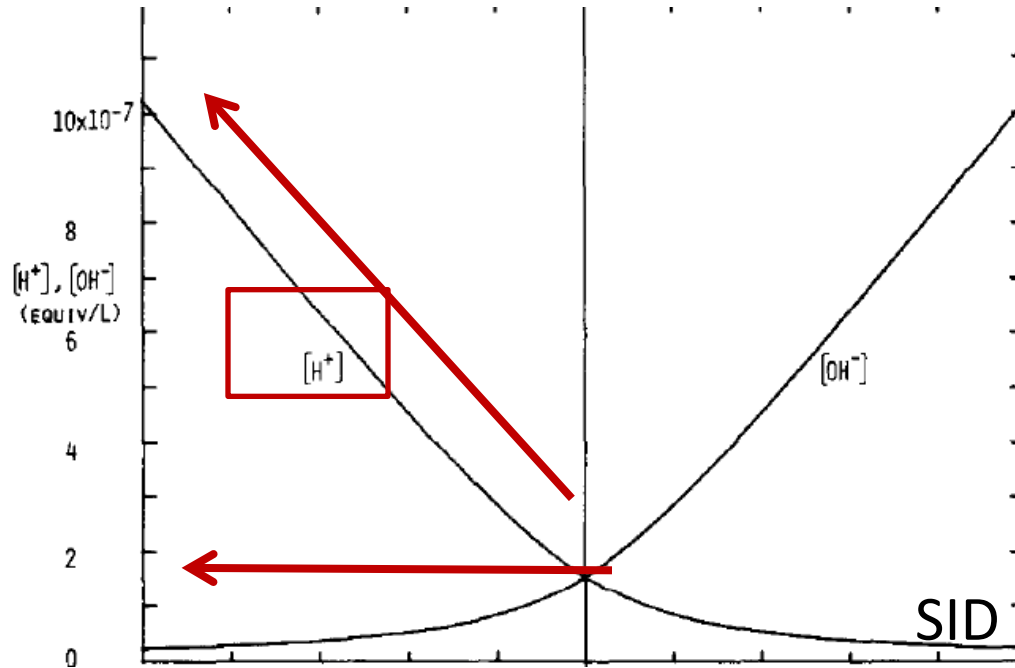
95/1110
8.6%

1% absolute mortality reduction – obviously not significant

dissolved CO_2 , H_2CO_3 , HCO_3^- , and CO_3^{2-} . Our three laws permit us to write eight simultaneous equations involving these components:

Water dissociation equilibrium:	$[\text{H}^+] \times [\text{OH}^-] = K'_w$
Weak acid dissociation equilibrium:	$[\text{H}^+] \times [\text{A}^-] = K_A \times [\text{HA}]$
Weak acid conservation of mass:	$[\text{HA}] + [\text{A}^-] = [\text{A}_{\text{TOT}}]$
CO_2 solution equilibrium:	$[\text{CO}_2 (\text{dsslvd})] = S_{\text{CO}_2} \times P_{\text{CO}_2}$
Carbonic acid equilibrium:	$[\text{H}_2\text{CO}_3] = S_{\text{H}_2\text{CO}_3} \times P_{\text{CO}_2}$
Bicarbonate equilibrium:	$[\text{H}^+] \times [\text{HCO}_3^-] = K_C \times P_{\text{CO}_2}$
Carbonate equilibrium:	$[\text{H}^+] \times [\text{CO}_3^{2-}] = K_3 \times [\text{HCO}_3^-]$
Electrical neutrality:	$[\text{S.I.D.}] + [\text{H}^+] - [\text{OH}^-] - [\text{A}^-] - [\text{HCO}_3^-] - [\text{CO}_3^{2-}] = 0$

What does a change in Cl^- do?



2. Alb^-
3. PaCO_2

1. $\text{Na}^+ - \text{Cl}^-$

$$\left. \begin{aligned} [\text{H}^+] &= \sqrt{K'w + ([\text{SID}]/2)^2} - ([\text{SID}]/2) \\ [\text{OH}^-] &= \sqrt{K'w + ([\text{SID}]/2)^2} + ([\text{SID}]/2) \end{aligned} \right\}$$

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Table S11. Indications for new renal replacement therapy.

	Balanced	Saline	
Indications for new RRT among patients who received new RRT, No. (%)	(n = 189)	(n = 220)	<i>P</i> value
Oliguria	144 (76.2)	180 (81.8)	0.16
Hyperkalemia with plasma potassium > 6.5 mEq/L	21 (11.1)	27 (12.3)	0.72
Acidemia with pH < 7.20	56 (29.6)	64 (29.1)	0.91
Blood urea nitrogen > 70 mg/dL	82 (43.4)	106 (48.2)	0.33
Plasma creatinine > 3.39 mg/dL	111 (58.7)	135 (61.4)	0.59
Organ edema	58 (30.7)	66 (30.0)	0.88
Other renal failure-related indication	19 (10.1)	21 (9.5)	0.86
Other non-renal failure-related indication	42 (22.2)	54 (24.5)	0.58